

BRIDGE COURSE BY SCHOOL OF ENGINEERING

UPES organizes Orientation and Induction program for the newly admitted students every year. The main objective of the Orientation program is to make the students aware of the academic aspects of the course, the rules and regulations of the University and ensuring active participation and progress of the students.

The Induction Program is thus designed to make the newly joined UG students feel comfortable, sensitize them towards exploring their academic interests and activities, promote bonding with each other, and build relations between teachers and students and to give a broader view of life.

The induction program includes Bridge Classes for Physics, Chemistry, Mathematics and Communication skills. It also includes various activities like Universal Human values workshop, visits to science regional centers, technology parks, museums, industries etc., First Aid workshop and counselling session, Social concern activity, Sports and Cultural activities, technical workshop, address by the spokespersons from Examination committee, Library, Discipline Committee, students' counselor, to mention a few.

1. Orientation

The purpose of Orientation on the very first day is to make all the newly admitted students acquainted with all activities organized in UPES for them, well in advance and it help them to choose an appropriate one to participate.

2. Familiarization with School/Department

Students are acquainted with their respective school/department/Program of study/laboratories/workshops/ICT facilities and other facilities. This interaction helps the students to differentiate between college life and school life along with career prospects offered by specific courses they have opted for.

3. Personality Enhancement Program (P.E.P.)

Personality Enhancement Program (P.E.P.) is an effort to update the base knowledge set of new students through intensive training experience for a smooth school to college transition. The balanced module helps the students to clearly communicate to collaborate with their colleagues after critical deliberations on their future course of action for next four years.

4. Physical Activities

Sports help an individual much more than in the physical aspects alone. It builds character, teaches and develop team spirit, strategic thinking, analytical thinking, leadership skills, goal setting and risk taking. The students enthusiastically participate in games of their interest.

5. Creative arts and culture program

Social concern activities aim to promote the students' interests and concern for social issues and to foster their passion for community service. This program also creates opportunities for the students to be engaged in social analysis and reflect on their experiences. The students prepare posters on various social themes and eventually come up with group presentations. The selected groups present a skit as well.

Along with this activity, 'Cultural Fiesta' provides a stage to the performers in the field of dance, singing, poetry recitation, acts. The students enthusiastically participate in the event, displaying their talent of singing, dancing (solo and group), poetry recitation, skit performance and comic acts.

6. Extracurricular activities

Extracurricular activities increase opportunities for social interaction and new relationship development. With this aim in mind, the activities are planned for various categories like Solo singing, group singing, poetry/comedy and dance, giving the students an exposure to perform before audience.

7. Universal Human Values Workshop

The goal of this workshop is to inculcate a deep sense of importance of core values in human life and ethics in society that students should live by. The program includes more interactive and experiential session so that the students learn to meditate and connect with their Higher Self within.

8. Visits to Local Area

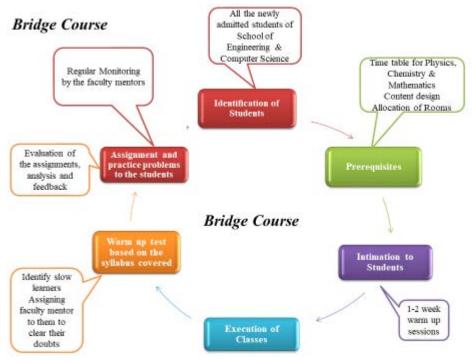
Visit to Regional Science & Historic Centers paves the way for the students to portray the growth of Science and Technology in our day to day life. The students are exposed to various science experiments relevant in day-to-day life, science museum, ecology of Uttarakhand, planetarium, 3D movie on environment, etc.

9. Bridge Course:

The refresher classes for Physics, Chemistry, Mathematics and Communication skills are organized in order to bridge the gap between their schooling and graduation. The main objective

University of Petroleum and Energy Studies

of the bridge course is to provide intellectual base to the students which eventually help them to face challenges of University education in a more efficient manner.



Sample case - School of Engineering conducted a Bridge Course from 31 July 2017- 5 August 2017. The details for the same are attached below

Dr. Piyush Kuchhal

From: Sent: To: Cc: Subject: Attachments: Dr. Piyush Kuchhal Friday, July 28, 2017 12:01 PM UPES-hod_ces Dr. Kamal Bansal; Dr. Manish Prateek; Dr. Suresh Kumar Warm-Up Classes for B. Tech Ist year students (Batch 2017-2018) Notice Group A&B.xlsx

Dear Colleagues,

It is to apprise you that the Department of Sciences is going to organize the warm-up classes followed by class test for newly admitted B. Tech Ist year students from July 31, 2017 to August 5, 2017. The purpose of a warm-up classes is to help the students revise their basic concepts and set the tone for what will follow. The class test would help us to customize the mode of instructions as per learning capability of the students.

The regular classes for B. Tech Ist year students will start from August 8, 2017.

The details of warm-up classes is attached with this mail for your ready reference and guidance to the students.

With regards,

Piyush Dua

NOTICE

WARM UP CLASSES

B.TECH 1st YEAR STUDENTS

31st JULY– 4th AUGUST This is to inform you that warm up classes for Physics, Chemistry & Mathematics will be conducted as per the schedule below:

BRANCHES	ROOM NUMBER	TIMING		
IFE (R680217001- R680217049)				
Aerospace Engg spl Avionics (R890217001-R890217025)	10101			
Aerospace Engg spl Avionics (R890217026- R890217042)				
Geo-Informatics Engineering (R620217001-R620217028)	10102			
Seo-Science Engineering				
Automotive Design Engg (R160217001-R160217069)	10103			
Aerospace Engineering (R290217001- R290217065)				
Power System Engineering (R630217001-R630217008)	10104	CHEMISTRY: 10.00 am - 11.30 am MATHEMATICS-11.30-1.30 pm LUNCH- 1.30pm-2.30pm		
Electronics & Communication Engg (R173217001-R173217057)	10105	PHYSICS- 2.30 pm- 4.30 pm		
APE+UP Stream (R870217002-R870217074)	10106			
APE+UP Stream (R870217075- R870217142)	10201			
APE+GAS Stream (R820217003- R820217072)	10202			
→PE+GAS Stream (R820217073-R820217139)	10203			
Chemical Engg R900217001-R900217067)	10204			
Chemical Engg R900217068-R900217137)	10205			

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(Dr. Piyush Kuchhal) Associate Dean, Apllied Sciences

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NOTICE WARM UP CLASSES B.TECH 1st YEAR STUDENTS 1st AUGUST- 4th AUGUST

This is to inform you that warm up classes for Physics, Chemistry & Mathematics will be conducted as per the schedule below:

BRANCHES	ROOM NUMBER	TIMING
Fire Safefty Engineering (R260217001- R260217057)		
Minning Engg. (R136217001-R136217010)	1201	
Mechatronics Engg (R880217001- R880217030)	1202	MATHMATATICS: 9:45 am-11:45pm PHYSICS-11.45-1.30 pm LUNCH- 1.30pm-2.30pm CHEMISTRY: 2.30 pm- 4.30 pm
Electrical Engineering(R132217001-R132217024)		
Mechanical Engg. (R113217001-R132217021)		
Mechanical Engg. (R113217022-R132217096)	1203	

(Dr. Piyush Kuchhal) Associate Dean, Apllied Sciences

SN	Faculty	Group	Room No.	TIMINGS
1	Dr. Maheshwar Pathak	Group-A1	10101	11:30-1:30PM
2	Dr. Monika Manglik	Group-A2	10102	11:30-1:30PM
3	Dr. Shweta Sachdeva	Group-A3	10103	11:30-1:30PM
4	Dr. Mithilesh Singh	Group-A4	10104	11:30-1:30PM
5	Dr. Anuj Kumar	Group-A5	10105	11:30-1:30PM
6	Dr. Sangeeta Pant	Group-A6	10106	11:30-1:30PM
7	Dr. Pradeep Malik	Group-A7	10201	11:30-1:30PM
8	Dr. Sanoj Kumar	Group-A8	10202	11:30-1:30PM
9	Dr. Mukesh Awasthi	Group-A9	10203	11:30-1:30PM
10	Dr. Sandeep Dixit	Group-A10	10204	11:30-1:30PM
11	Dr. Girish Dobhal	Group-A11	10205	11:30-1:30PM

CNI	Faculty	GROUP	Room No.	TIMINGS
SN	and a second	Group-B1	1201	9:45AM-11:45AM
	Dr. Manoj Kumar	Group-B2	1202	9:45AM-11:45AM
	Dr. Mradul Veer Singh	Group-B3	1203	9:45AM-11:45AM
3	Dr. Shashank Chaube	51000		

MATH 1001-GROUP B

	MATH1002-GROUP C	
SN	Faculty	Group
1	Dr. Shashank Chaube	Group-C1
2	Dr. Pratibha Joshi	Group-C2
3	Dr. Anupam Bhandari	Group-C3
4	Dr. Mradul Veer Singh	Group-C4
5	Dr. Reshu Gupta	Group-C5
6	Dr. Pankaj Kumar Mishra	Group-C6
7	Dr. Pavan Kumar Pannala	Group-C7
8	Ravi Kiran Maddali	Group-C8
9	Dr. Akmal Husain	Group-C9
10	Dr. Anurag Shukla	Group-C10
11	Dr. Nitin Uniyal	Group-C11
12	Dr. Manoj Kumar	Group-C12
13	Dr. Mithilesh Singh	Group-C13
14	Dr. Pradeep Malik	Group-C14
15	Dr. Sangeeta Pant	Group-C15

BRIDGE COURSE (FOR NON CIT)

- **1. MATRICES-** Types of matrices: Symmetric, Skew-symmetric, Orthogonal, Idempotent and Involutory, Determinant and its properties, Elementary row transformations, Solution of system of linear equation.
- 2. DIFFERENTIAL CALCULUS- Limits and Continuity, Differentiability and Monotonicity, Derivative of a function, Tangent and Normal, Maxima and Minima, Mean value theorems.

2 hrs.

2 hrs.

3. INTEGRAL CALCULUS- Integration and standard results, Integration by substitution, Integration by parts, Integration by partial fraction, Definite integral and its properties, Area under curve, Area between two curves.

2 hrs.

4. DIFFERENTIAL EQUATION- Order and Degree of differential equation, Solution of first order first degree differential equation, Solution of first order homogeneous differential equation.

1 hr.

5. TRIGNOMETRY & COMPLEX NUMBERS- Trigonometric identities, Euler's formula, De Moivre theorem, Cartesian plane, Polar plane, nth root of unity

1 hr.

6. GEOMETRY- Straight line, Pair of straight line, Circle, Parabola, Ellipse, Hyperbola, Direction Cosine, Direction ratio, Plane.

1 hr.

7. VECTOR & PROBABILITY- Definition of vectors, Addition of vectors, Dot product, Cross product, Permutation and Combination, Basic probability theory.

1 hr.

BRIDGE COURSE (FOR CIT)

- **1. SET THEORY-** Sets and their operations (Union, Intersection, Complement, Difference and symmetric difference), Cartesian product of sets and properties. *1 hr.*
- 2. DIFFERENTIAL CALCULUS- Limits and Continuity, Differentiability and Monotonicity, Derivative of a function 1 hr.
- **3. INTEGRAL CALCULUS-** Integration and standard results, Integration by substitution, Integration by parts, Integration by partial fraction, Definite integral and its properties, Area under curve, Area between two curves.

2 hrs.

 MATRICES- Determinant and its properties, Some special matrices like Symmetric, Skew-symmetric, Orthogonal, Idempotent and Involutory, Elementary row transformations. Dr. Maheshwar Pathala AI

S.NO.	BRANCH	ROLL NUMBER	SAP ID	NAME	ROOM NO.	3/07/17	01/08/17	02/08/17	3/08/17	H/08/17
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3	IFE 2017	R680217003	500062740	ADITI MISHRA	10101	ALIE	Auto	Adita		
4	IFE 2017	R680217004	500061618	ADITYA RAJ	10101	Aditya	Addys	Addya	Attys	Aditya
5	IFE 2017	R680217005	500062252	ANANT KUMAR SINGH	10101	Anont.	Ament:	Anant.	Anant	Anost
6	IFE 2017	R680217006	500063115	ANUJ JEENA	10101	Ang	Any			
7	IFE 2017	R680217007	500060172	ANURAG .	10101	Anutag	Anyras	Anyrag	Amuray	Anyrag
8	IFE 2017	R680217008	500062631	ARJUN SHARMA	10101	Har	Sharp-	- Albertin	harry	E
9	IFE 2017	R680217009	500060229	ASHISH PADIYAR	10101	- gading	By th	Ater	ADD	Prov.
10	IFE 2017	R680217010	500062951	ASHUTOSH DUBEY	10101	Shutost	Forther.	Apph	Ashit	Aura
11	IFE 2017	R680217012	500062546	AYUSH SINGH	10101	Arigo	sand	assing	pyus	
12	IFE 2017	R680217013	500062771	* BIPUL KUMAR	10101	AA 11-	Bipyth	AN	Deputation	Bipuelo
13	IFE 2017	R680217014	500060158	CHETAN AGRAWAL	10101	hit	CAND	Com	CAR	0
14	IFE 2017	R680217015	500061155	DHRUV AHUJA	10101	otherenth	John	ASUL	affect	aller
15	IFE 2017	R680217016	500062939	HARSHIT GAUTAM *	10101	Harshit	Houshi	Houst	partiel	Harry
16	IFE 2017	R680217017	500060005	HIMANSHU KUMAR	10101	Himawan	Himada	Stimate	Timantes	Plinamida
17	IFE 2017	R680217018	500063396	HRITIK SINGH	10101	with	Hitch	fortik	thit	
18	IFE 2017	R680217019	500061024	ISHANK MISHRA	10101	FUANT	ANANK .	FURNE	5HANK	FARE
19	IFE 2017	R680217021	500062042	JAYESH PRASAD	10101	Jaye	Layy	rayot	Jacon	
20	IFE 2017	R680217022	500060153	KARAN DHAWAI	10101	Ranga	Donan	Coroll	(A) roy	
21	IFE 2017	R680217023	500062779	MONTY GOYAL	10101	Monty	Honty	Nonty	Monty	. Monthy
22	IFE 2017	R680217024	500062564	NAMAN SOLANKI	10101	aslant	appline	Nietoki.		
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25	IFE 2017	R680217029	500063057	PRIYANK SHARMA	10101					
26	IFE 2017	R680217030	500063086	RINKLE BHADANA		Sont.	Recent	Eningh.	Dama	-
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University of Petroleum & Energy Studies

College of Engineering Studies

Warm Up Test-5th August, 2017

Name of Student:	Branch:	
Enrollment No	SAP ID	
Time: 2 hours 15 minutes	Max. Marks: 80	

Do not open this booklet until you are told to do so by the invigilator.

Read the instructions carefully before you start the exam

- 1. The question paper has separate sections for Physics, Chemistry and Mathematics with a total of 80 questions.
- 2. Before you start the examination, check that your answer sheet is free from printing defects, i.e. faded print, missing print, repetitive defects etc.. Ask the invigilator to replace your answer sheet if it has printing defects.
- 3. All the questions are multiple choice questions with 4 options: A, B, C, D.
- 4. Read the questions carefully and choose the ONE best answer.
- 5. Mark only one answer. Multiple answers will be cancelled.
- 6. Use only pen to fill the answer sheet.
- 7. Each guestion carries 1 mark for each correct answer.
- 8. No negative marking.
- 9. Do NOT fold or crease your answer sheet.
- 10. Make sure that all the entries are filled correctly on the booklet as well as answer sheet.
- 11. At the end of the exam, answer sheet as well as the question booklet has to be handed over to the invigilator.
- 12. No student is allowed to submit the answer sheet and leave the room before 1 hour of the commencement of exam.

PHYSICS

- 1. Calculate the number of photons, from green light of mercury ($\lambda = 4961$ Å), required to do one joule of work. (A) 4524.2×10^{18} (B) 2.4961×10^{18} (C) 2.4961 (D) 2.4961
- 2. In which of the following the interference is produced by the division of wavefront(A) Michelson's interferometer
 - (B) Soap bubble
 - (C) Newton's ring
 - (D) Biprism
- 3. In Young's double slit experiment the bright and dark fringes are obtained on a screen. If the screen is gradually moved towards the slits, then it is observed that the separation between the successive fringes
 - (A) Increases (B) decreases
- (C) remain constant (D) none of these

4. When a thin transparent sheet is placed in the path of one of the interfering waves then

(A) The entire fringe system shift towards the side on which the sheet is placed

(B) The entire fringe system shift towards the opposite side on which the sheet is placed(C) There is no shift in the fringe pattern

(D) the fringe system becomes narrower

5. Which of the following is conserved when light waves interfere?

- (A) Intensity (B) energy
- (C) amplitude (D) momentum
- 6. A light wave can travel
 - (A) In vacuum
 - (B) in material medium
 - (C) in both
 - (D) none of these
- 7. When light is refracted, which of the following does not change?
 - (A) Wavelength (B) velocity
 - (C) frequency (D) amplitude
- 8. As per Gauss's Law, the total electric flux φ through a closed surface and the total charge q enclosed by that surface are related as
 (A) Ø=∫Bds
 (B) Q=Ø
 (C) Both
 (D) None

9. Which of the following is true for electrostatics?.

(A) $E = -\nabla V$	(B) $\nabla^2 = 0$
(C) V=−∇E	(D) None of these

10. Electric flux density_____medium.

- (A) depends on the
- (B) is independent of the
- (C) both (A) and (B) (
- (D) None of these

11. A charge of 12 C has velocity of 5a_x + 2a_y - 3a_z m/s. Determine F on the charge in the field of (i) E=18a_x,+5a_y +10a_z V/m (ii) B = 4a_x + 4a_y + 3a_z wb/m².

(A) F = 254.27 N and 415.17 N

(B) 154.17 N and 315.17 N

(C) Both are possible

(D) Nil

12. Magnetic flux density is a relation of

(A) Current and area

(B) Area and its direction

(C) Magnetic flux and area

(D) None

13. Magnitude of electric dipole moment of a dipole with charges q separated by a distance d is given as

(A) q/d	(B) qd
(C) d/q	(D) None of these

14. Curl of electrostatic field is

(A)	00	(B)0

(C) 1 (D) None of these

15. $F=qB \times V$ represents the force exerted on a

- (A) Charge q moving with velocity V in an electric field
- (B) Charge q moving with velocity V in a magnetic field B

(C) Both

(D) None

16. The magnitude of the emf induced is directly proportional to rate of change of magnetic flux. This is put forth by

(A) Bio-Savart's law (B) Faraday's I law

(C) Ampere's law (D) Faraday's II law

- 17. Curl of magnetic field intensity is
 - (A) current density
 - (B) magnetic flux density
 - (C) current
 - (D) zero
- 18. Total flux passing through a closed surface held in a magnetic field is
 - (A)∞ (B)0
 - (C) 1 (D) None
- 19. When two similar metal balls with charges of +2C and -0.5C are brought in contact and then separated, the charges on each metal ball will be
 - (A) Equal and same
 - (B) unequal and same
 - (C) equal and opposite
 - (D) Unequal and opposite
- 20. Magnetic monopoles does not exist, may be contained in
 - (A) Gauss law of magnetism
 - (B) Biot Savert's law
 - (C) Ampere's law
 - (D) Faradey's law

21. Maxwell's equations are based on law(s)

14 / (3).	
(A) Faraday's	(B) Gauss's
(C) Ampere's	(D) All of these

22. Displacement current density J_D is equal to

(A) $\delta D/\delta t$	(B) 0
(C) 1	(D) - δD/δt

23. A car with speed of 22.2 m/s blowing it's horn while it is approaching a dog sitting on the road. If the frequency of the horn is 437.9 Hz and the speed of sound is 337 m/s, what might be the frequency that dog hears?

(A) 330 Hz	(B) 468Hz
(C) 685 Hz	(D) 440Hz

24. 90 dB is _____ more sound pressure than 70 dB

(A) 100 times	(B) 20 times
(C) 10 times	(D) 44times

25. Sound is produced due to

(A) friction	(B) circulation
(C) vibrations	(D) refraction

26. In which of the following, the speed of sound is least?

(A) Air	(B) Liquid
(C) Solid	(D) Vaccum

27. Wavelength of ultrasonic waves is

(A) more than audible sound

(B) less than audible sound

(C) equal to audible sound

(D) greater than speed of light

28. Reflection of sound is called (A) quality (B) loudness (C) pitch (D) echo

29. A human ear can hear sounds in range of

(A) 40-40,000 Hz (B) 30-35 Hz (C) 20-20000 Hz (D) 50-50000 Hz 30. Technique used to absorb noise by using soft and porous surfaces is called

(A) acoustic protection

(B) refraction

(C) absorption

(D) semi lunar protection

CHEMISTRY

- 31. Calculate Enthalpy of formation of Methane (in KCal/mol), if its enthalpy of combustion is -890.35 KJ/mol and enthalpy of formation of water and carbon dioxide is -285.84 Kj/mol and -393.51 Kj/mol respectively.
 - (A)-74.84
 - (B) -17.81
 - (C) -20.01
 - (D)-223.5
- 32. Fractional distillation is used when difference in
 - (A)Melting points of two solids is less.
 - (B) Boiling points of two liquids is large.
 - (C) Boiling points of two liquids is small.
 - (D) Melting points of two liquids is large.
- 33. Determine ΔE for below mentioned chemical transformation, if ΔH_f° for CO₂ (g), CO (g) and H₂O (g) are 393.5, -111.31 and -241.8 kJmol⁻¹ respectively at 298 K

 $\operatorname{CO}_2(g) + \operatorname{H}_2(g) \rightarrow \operatorname{CO}(g) + \operatorname{H}_2\operatorname{O}(g)$

- (A)40.39 kJ/mol
- (B) -40.39 kJ/mol
- (C) 746.61 kJ/mol
- (D)-746.61 kJ/mol

- 34. Which one of the following statements best describes the enthalpy change of a reaction?
 - (A) The energy released when chemical bonds are formed during a chemical reaction
 - (B) The energy consumed when chemical bonds are broken during a chemical reaction
 - (C) The difference between the energy released by bond formation and the energy consumed by bond cleavage during a chemical reaction
 - (D) The increase in disorder of the system as a reaction proceeds
- 35. The enthalpy of combustion of galactose is -2908 kJ mol⁻¹. Which one of the following statements regarding this process is false?
 - (A) The products of the combustion of galactose are less stable than galactose itself
 - (B) This process is exothermic
 - (C) Overall, energy is liberated when this process happens

(D) None of the above

- 36. The EMF of the electrochemical cell, Fe, Fe²⁺ (0.1M) | X²⁺ (0.001M), X; will be?
 If E^o (Fe²⁺ /Fe) = 0.44V and E^o (X^{2+/}X) = -0.40V
 (A) -0.889 V
 (B) +0.889 V
 (C) -1.501 V
 - (D)+1.501V

37. An electrochemical cell is also called

(A) battery cell

(B) galvanic cell

(C) chargeable cell

(D)None of the above

 Potential of single hydrogen electrode is considered as

(A) zero

(B) unity

(C) constant

(D) multiple of 1

39. The electrical resistance of a column of 0.5 M KOH solution having diameter is 1 cm and length 78.5 cm is 5000 ohms. The resistance of the above solution is?

(A) 20 ohm

- (B) 40 ohm
- (C) 30 ohm

(D) 50 ohm

- 40. The element that act as anode always has -----value of reduction potential in the ECS
 - (A) higher
 - (B) lower
 - (C) in middle

(D) no effect of position

41. The shape of $C_2H_5^+$ is

(A) Triangular planar

(B) Square planar

(C) Tetrahedral

(D) None of the above

42. Which one of the following is the weakest acid?

(A) CH₃COOH

(B) CCl₃COOH

(C) C - l₂COOH

- (D) CH₃CH₂COOH
- 43. Identify the false statement regarding resonance
 - (A)As the number of resonating structures increases, the structure becomes more stable.
 - (B) Structure having zero degree of resonance is the most stable one
 - (C) Atoms with full octet resonance form are more stable when compared with the one with unfilled octet
 - (D) A and C are correct
- 44. Select the incorrect statement:
 - (A) A resonance may sometimes cause sp³ atoms to become sp² hybridized
 - (B) Delocalizing one lone pair causes aromaticity
 - (C) one lone pair will be counted as two pi electrons according to Huckel's equation
 - (D)Two sigma bonds make up a double bond

45. Which one the following concept is temporary in nature

- (A) Electromeric
- (B) Resonance
- (C) Hyperconjugation
- (D) Inductive

- 46. For a zero order reaction
 - (A) The rate of reaction is zero
 - (B) The rate of reaction is independent of the concentration of reactant
 - (C) The rate constant of the reaction is zero
 - (D) None of the above
- 47. The role of catalyst is-
 - (A)To bring the chemical reaction towards equilibrium
 - (B) To Decrease the activation energy of the chemical reaction
 - (C) To provide an alternate path, mechanism of the catalyzed reaction
 - (D) All of the above
- 48. The rate law of a reaction is given by; rate = k [NO]² [O₂], When the initial concentration of NO is tripled, the rate will be?
 - (A) Decreases by a factor of nine
 - (B) Increases by a factor of three
 - (C) Increases by a factor of six
 - (D) Increases by a factor of nine
- 49. A straight line graph will be obtained for a first order reaction having negative slop, if we plot a graph between-
 - (A) concentration of reactant and time
 - (B) inverse of concentration of reactant and time
 - (C) log of concentration of reactant and time
 - (D) none of the above

- 50. The reaction 2A→Product, follows zero order reaction kinetics, when the initial concentration of A is 0.5M and the half-life period is 8 minutes. What will be the half-life period, if the initial concentration of A is 0.1M?
 - (A) 16 minutes
 - (B) 8 minutes
 - (C) 40 minutes
 - (D) 1.6 minutes
- 51. Which of the following statements is false?
 - (A) The repeat unit in natural rubber is isoprene.
 - (B) Both starch and cellulose are polymers of glucose.
 - (C) Artificial silk is derived from cellulose.
 - (D)Nylon-66 is an example of elastomer
- 52. Which polymers occur naturally?
 - (A) Starch and Nylon
 - (B) Starch and Cellulose
 - (C) Proteins and Nylon
 - (D) Proteins and PVC
- 53. Which of the following is a fully fluorinated polymer?
 - (A) Neoprene
 - (B) Teflon
 - (C) Thiokol
 - (D)PVA

- 54. Polymer formation from monomers starts by
 - (A) condensation reaction between monomers
 - (B) coordination reaction between monomers
 - (C) conversion of monomers to monomer ion by protons
 - (D) hydrolysis of monomers
- 55. Which of the following is a polyamide?
 - (A) Teflon
 - (B) Nylon-6,6
 - (C) Terylene
 - (D) Bakelite

MATHEMATICS

56. Let f(x) = |x| + |x - 1|; then (A) f(x) is continuous both at x = 0 and 1. (B) f(x) is continuous at x = 0 but not at x = 1.(C) f(x) is continuous at x = 1 but not at x = 0.(D) None of these 57. The function $f(x) = |\sin x|$ is (A) Continuous and Differentiable. (B) Discontinuous. (C) Continuous but not Differentiable. (D) None of these 58. If f(x) = |x - 2| and g(x) = f(f(x)), then g'(x) for x > 2 is (A)1. (B) 2.

(C)-1. (D) None of these

59. The curve given by $x + y = e^{xy}$ has a tangent parallel to the y-axis at the point (A)(0,1). (B) (1,1). (C)(1,0). (D) (0,0).

60. Maximum value of $f(x) = x^3 + x^2 - x + 1$ on the interval [-2, 1/2] (A)2. (B) -1. (C)22/27. (D)5.

61. Integral $\int_0^2 \sin \frac{\pi[x]}{2} dx$ ([x] represent greatest integer function of x) is (A)0. (B)-1. (C)1. (D)None of these.

62. The area bounded by the curve y = x|x|, x-axis and the ordinates x=1, x=-1 is given by (A)0. (B) 1/3. (C)2/3. (D)None of these. 63. Solution of the differential

equation xdy - ydx = 0 represents

(A)Parabola whose vertex is at origin

- (B) Circle whose center is at origin.
- (C)A rectangular hyperbola.

(D)Straight line passing through origin.

64. The inverse of a skew symmetric matrix of odd order is

- (A) A symmetric matrix.
- (B) A skew symmetric matrix.
- (C) Diagonal matrix.
- (D) Does not exist.

65. If A is 2×2 matrix such that $A^2 = 0$, then trace(A) is (A)1. (B) -1.

(C)0. (D)None of these.

66. If A is a 4×4 matrix such that |A| = 4, find |adj A|(A)16. (B) 32. (C)128. (D)64.

67. If the system of equations

x + 2ay + az = 0x + 3by + bz = 0x + 4cy + cz = 0

has a nontrivial solution, then a, b, c(A) satisfy a + 2b + 3c = 0.

- (B) are in A.P.
- (C) are in G.P.
- (D) are in H.P.

68. The range of the function

$$f(x) = x^2 + \frac{1}{x^2 + 1}$$

is

(A) $[1,\infty)$. (B) $[2,\infty)$. (C) $[3/2,\infty)$. (D)None of these.

1

69. There are 2 teams with n persons in each. be seated in a row so that they are alternate The probability of selecting 2 persons from one team and 1 person from the other team is 6/7. Then n=

(A)3.	(B) 4.
(C)5.	(D)6.

70. Let a set of first 20 natural number is given, the probability if selected number from the set is squared and ends in six is

(B)2/5.(A)1/5.(C)3/20. (D)3/5.

71. Using De Moivre's theorem, write $(1+i)^4$ in the form of a + ib

(A) - 4.(B)4. (C) - 4 + 4i.(D)4 - 4i.

72. If ω is a cube root of unity, then roots of the equation $(x-1)^3 + 8 = 0$, are $(A)-1, 1-2\omega, 1-2\omega^2.$ (B)-1, $1+2\omega$, $1+2\omega^2$. (C)-1, $-1 + 2\omega$, $-1 - 2\omega^2$. (D)-1, -1, -1.

73. If ω is a cube root of unity, such that $n \in N, \ \omega \neq 1$, then $\omega^{3n+1} + \omega^{3n+3} + \omega^{3n+5}$ equals (A) 0. (B)1. (C)-1. (D)3.

74. Find	the value of
$\sin\left(\frac{\pi}{2}-\right)$	$\sin^{-1}\left(\frac{-1}{2}\right)$
(A)1/2. (C)1/4.	(B)1/3.
(C)1/4.	(D)1.

75. If the acute angle between the lines x +ky + 3 = 0 and 2x + y - 7 = 0 is $\tan^{-1}(3/4)$, then k is (A)1. (B)2. (D)4. (C)3.

76. A point inside the circle $x^2 + y^2 + 3x - x^2 + y^2 + y^2$ 3y + 2 = 0 is (A)(-1, 3).(B)(-2, 1).(C)(2, 1).(D)(-3, 2).77. In how many ways 4 boys and 3 girls can

(B)288. (A)144. (C)12. (D)256.

78. Flat surface in which two points are joined by using straight line is classified as

(A)Line. (B)Ray.

(C)Plane. (D)Intersecting Line.

79. Find the vector $u \times v$ where u = [3, , -1, 1]and v = [2, 5, 1](A)[-6, -1, 17].(B)[-6, 1, 17].(C)[4, 5, 13]. (D)[-4, -1, 15].

80. Find non zero scalars α , β such that for every vectors a and b $\alpha(a+2b) - \beta a + (4b-a) = 0$ (A) $\alpha = 2, \beta = 1.$ (B) $\alpha = -2, \beta = -3.$ (C) $\alpha = 1, \beta = 3.$ (D) $\alpha = -2, \beta = 3.$

2

Student N					Answer She		and the second se	
seature 1	lame:					SAP ID:	5000	
Branch:						Room No:		
Enrollme		R				Date:	05.	Aug.2017
Q.No	Answer	Marks	Q.No	Answer	Marks	Q.No	Answer	Marks
Q1	B	Sec-1:	Q31	B	Sec-5:	Q56	A	Sec-10:
Q2	D		Q32	C		Q57	C	
Q3	B		Q33	A		Q58	A	
Q4	A		Q34	C		Q59	C	
Q5	B		Q35	A		Q60	A	
Q6	C		Q36	A	Sec-6:	Q61	C	
Q7	C	00	Q37	B		Q62	C	
Q8	A	Sec-2:	Q38	A		Q63	D	*
Q9	A		Q39	D	and the second sec	Q64	D	Sec-11:
Q10	B		Q40	B		Q65	Ć	
Q11	A		Q41	A	Sec-7:	Q66	D	
Q12	C		Q42	P		Q67	D	
Q13	B		Q43	B		Q68	A	
Q14	B		Q44	D		Q69	B	
Q15	B	0.0	Q45	A		Q70	A	
Q16	B	Sec-3:	Q46	B	Sec-8:	Q71	A	Sec-12:
Q17	B		Q47	D		Q72	A	
	B		Q48	D		Q73	A	
Q19	A		Q49	C		Q74	A	_
Q20	B		Q50	C		Q75	B	Sec-13:
Q21	0		Q51	B	Sec-9:	Q76	B	
	4	See 4:	Q52	B		Q77	A	
	B	Sec-4:	Q53	B		Q78	C	
	A		Q54	A	-	Q79	A	
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MATH-1001

No	Roll No	SAP ID	Name of the Student	Branch	Mathematics-Total Marks	Name of the Student Mentor
	R160217027		JAYESH SETH	ADE 2017	6	
1	D160217027	500060686	MOHIT AGRAWAL	ADE 2017	6	
	R160217034 R820217021	500061003	ANIMESH KRISHNAN	APE GAS 2017	2	
	R820217021 R820217067	500061452	PIYUSH PRADHAN	APE GAS 2017	6	Dr Anurag Shukla
4	R820217007	500061773	PUSHPESH RANJAN	APE GAS 2017	6	
	R820217078	500060021	VIKRANT THAKUR	APE GAS 2017	6	
6	R820217131	50006021	VISHESH SHARMA	APE GAS 2017	6	
1	R820217133	500060864	ARJIT AGARWAL	APE UP 2017	0	
8	R870217037	500060063	ASHUTOSH RAWAT	APE UP 2017	0	
9	R870217044	500060883	BALRAM CHOUDHARY	APE UP 2017	0	Dr Maheshwar Pathak
10	R8/021/048	500060872	DARPAN SINGH	APE UP 2017	0	
10	R8/0217051	500060675	HARSH KATIYAR	APE UP 2017	0	
12	R870217001	500060015	KUMAR VATSAL	APE UP 2017	0	
13	R8/0217073	500060810	MANAS BHASIN	APE UP 2017	0	
14	R870217075	500060100	MANAV KUMAR	APE UP 2017	0	
10	D970217070	500060208	NEETHESH JAYARAMAN	APE UP 2017	4	Dr Anuj Kumar
10	R 200217012	500062363	GHANSHYAM RAI	ASE 2017	6	
1.0	R290217012	500062505	TEJASHVINI BATHEJA	ASE 2017	5	
10	R290217037	500063804	SUNIDHI BATRA	ASE 2017	6	
			ANURAG RANA	ASE+AVE 2017	5	Dr Mrinal Jana
			ISHAN SINGH	ASE+AVE 2017	5	
2	R890217036	500062491	SATYA DASH	ASE+AVE 2017	6	
2.	2 R900217013	50000224	AMBUJ TIWARI	Chemical 2017	4	
2.	1 P000217016	50006144	ANSHIKA ANNU	Chemical 2017	6	
2	4 R900217104	50006021	SHIVAM SAMANT	Chemical 2017	3	
2	5 R680217016	50006021	HARSHIT GAUTAM	Civil 2017	6	Dr Shweta Sachdeva
2	6 R680217017	50006293	LIIMANCHI VIMAP	Civil 2017	6	
2	/ K68021704	50006000	5 HIMANSHU KUMAR	Civil 2017	5	
2	8 R68021704	50006217	8 UTKARSH ARUN	Civil 2017	6	
2	9 R173217009	50006242	9 YASH SONI	ECE 2017	1	
3	0 R26021700	50006242	6 DEVADEEP CHAKRABORTY	FSE 2017	6	
3	1 R26021704	5 50006253	AASHI DESHPANDE	FSE 2017	5	Dr Monika Manglik
3	2 R62021701	7 50006196	6 SUDHANSHU GUPTA	GIE 2017	6	
			9 NEETI MAHAJAN	GSE 2017	5	
3	4 R49021702	3 50006293	4 APURV PREM	GSE 2017 GSE 2017	6	
3	5 R11321700	5 50006306	4 SK ARBAZ ALI	Mechanical 2017	6	
			8 AKHIL BHARDWAJ	Mechanical 2017 Mechanical 2017	6	Dr Divya Ahluwalia
3	7 R11321704	2 50006298	7 JAI KUMAR	Mechanical 2017 Mechanical 2017	6	Dr. Dr. ju. Minardana
3	8 R11321704	8 50006285	1 MANSI SRIVASTAVA	Mechanical 2017 Mechanical 2017	5	-
3	9 R88021700	2 50006152	8 NINAD PANCHAL	Mechatronics 2017	5	
			3 ABHINAV MUDGAL		5	Dr S K Banerjee
4	1	50006299	5 SANJOY GHOSH	Mining 2017	2	DI SIX Duncijec

Slow learners Identified on the basis of Maribs luss than . 30%. Mentors assigned to them

MATH 1002

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	2 R103217028 500063517 AYUSHI GUSAIN	CS BAO 2017	5	
	3 R103217032 500063558 DEEPAK .	CS BAO 2017	5	
	4 R103217035 500062828 DHANANJAY KHAJURIA	CS BAO 2017	5	-
Ē	5 R103217043 500062924 GUDDY KUMARI	CS BAO 2017	5	Dr Shashank Chaube
	6 R103217055 500062393 KOMALPREET KAUR	CS BAO 2017	5	
	7 R103217056 500062707 KSHITIJ JINDAL	CS BAO 2017	5	
-	8 R103217059 500061589 MADHUR RAKHEJA	CS BAO 2017	5	
Γ	9 R103217064 500062965 NAMRATA AGARWAL	CS BAO 2017	5	
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F	14 R103217083 500062971 RHITIK	CS BAO 2017	5	
T	15 R103217091 500062845 SAMANTA AGARVAL	CS BAO 2017	5	Dr Reshu Gupta
F	16 R103217096 500062592 SHAKSHAM SISODIA	CS BAO 2017	3	D. House Oupen
F	17 R103217100 500063082 SHIRAZ HUSSAIN	CS BAO 2017	5	
	18 R103217101 500060074 SHIVAM KUMAR	CS BAO 2017	3	-
	19 R103217102 500063089 SHIVANGI SHARMA	CS BAO 2017	5	-
F	20 R103217105 500062952 SHUBHANKAR JAISWAL	CS BAO 2017	5	
F	21 R103217106 500063452 SHWETA RAWAT	CS BAO 2017	4	
F	22 R103217109 500062397 SOMYA SHARMA	CS BAO 2017	5	Dr Akmal Husain
F	23 R103217110 500063445 SONA JAIN	CS BAO 2017	5	Di Akinai Husaili
F	24 R103217111 500062738 SUN GAJIWALA	CS BAO 2017	3	
-	25 R103217120 500062545 TWINKLE BHALLA	CS BAO 2017	4	-
-	26 R103217121 500061932 UDISHA KUMAR	CS BAO 2017	2	-
-	27 R103217124 500062787 VANSHIKA SINGHAL	CS BAO 2017	5	Dr Anuman Dhandard
-	28 R103217127 500063330 VEDANT DARUKA	CS BAO 2017	5	Dr Anupam Bhandari
-	29 R103217128 500062532 VEDANT MITTAL	CS BAO 2017		-
-	30 R103217120 50002552 VEDANT MITTAL 30 R103217130 500062523 VIDIT NANU		5	_
-	31 R103217133 500060079 VISHESH GOEL	CS BAO 2017 CS BAO 2017	5	B.B.J. MUD
-	32 R103217135 500060475 VISILESH GOEL	CS BAO 2017 CS BAO 2017	5	Dr Pradeep Malik
-	33 R103217141 500063076 HARSHITA GUPTA		5	-
-	34 R133217005 500062514 HARSH DAHIYA	CS BAO 2017	5	-
-	35 R133217005 500063017 MINAL SAHU	CS BFSI 2017	4	-
-		CS BFSI 2017	5	
-	36 R133217012 500062782 SANYA CHHABRA	CS BFSI 2017	3	
-	37 R133217013 500062334 SAUMYA AHUJA	CS BFSI 2017	5	
-	38 R133217014 500062282 SAUMYA SRIVASTAVA	CS BFSI 2017	5	
-	39 R172217016 500061930 AYUSH KUMAR	CS Bigdata 2017	2	Dr Pankaj Kumar Mishra
-	40 R172217025 500060242 KARTIKAY BANSAL	CS Bigdata 2017	5	_
-	41 R172217026 500061543 KOMPAL SITHTA	CS Bigdata 2017	4	_
H	42 R172217042 500061994 RITIKA SHARMA	CS Bigdata 2017	4	_
-	43 R172217056 500061371 URVEE KUMAR	CS Bigdata 2017	5	
-	44 R172217059 500060257 VASUNDHARA GARG	CS Bigdata 2017	5	Dr Girish Dobhal
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-	46 R110217103 500060041 NIKITA KUKRETI	CS CCVT 2017	5	
L	47 R110217104 500061707 NISHANT SHUKLA	CS CCVT 2017	4	
L	48 R110217112 500061057 PRASHANSA GUPTA	CS CCVT 2017	4	-
L	49 R110217130 500061454 S MRINAL	CS CCVT 2017	4	
-	50 R110217148 500061171 SHIKHAR SHUKLA	CS CCVT 2017	. 3	
L	51 R110217191 500061708 ANMOL SRIVASTAVA	CS CCVT 2017	5	
	52 R134217011 500061933 ADRISH MITRA	CS CSF 2017	5	Dr Mithilesh Singh
L	53 R134217013 500062017 AKHIL SINGH	CS CSF 2017	4	_
L	54 R134217021 500061517 AMAN JAIN	CS CSF 2017	5	
	55 R134217024 500061409 ANADI PANDEY	CS CSF 2017	.5	
	56 R134217069 500060161 KARTIK CHAUHAN	CS CSF 2017	5	
L	57 R134217081 500061643 MANSI BISHT	CS CSF 2017	4	
L	58 R134217087 500062101 MOHIT YADAV	CS CSF 2017	5	
L	59 R134217106 500061997 PRANAY SURI	CS CSF 2017	4	Dr Sandeep Dixit
L	60 R134217108 500061936 PRANJAL SINGH	CS CSF 2017	5	
L	61 R134217109 500060027 PRASHANT SINGH	CS CSF 2017	.5	
Γ	62 R134217120 500062106 RISHAV SHARMA	CS CSF 2017	5	
	63 R134217128 500061904 ROMIL VERMA	CS CSF 2017	5	
Г	64 R134217135 500061056 SAKSHAM BHUTANI	CS CSF 2017	5	
F	65 R134217161 500061906 SOMYA AGRAWAL	CS CSF 2017	5	
F	66 R134217184 500061992 VIVEK JAISWAL	CS CSF 2017	1	Dr Mukesh Kumar Awasthi
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1	69 R171217017 500060134 GARISHMA VIRK	CS DevOps 2017	5	
F	70 R171217028 500062098 KESHAV MISHRA	CS DevOps 2017 CS DevOps 2017	5	-
£	71 R171217049 500060009 RACHIT CHAUHAN	CS DevOps 2017	5	-

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77 R164217019 500	0060306 HARDIKI SINGHAL	CS 1oT 2017	5	
78 R164217025 500	0062727 KARTIKEY SHRINGI	CS IoT 2017	5	
	0063385 MAHAK GUPTA	CS IoT 2017	5	
80 R164217031 500	062087 NANDAN BHARTI	CS IoT 2017	4	Dr Sanoj Kumar
	0062280 NIHIT GARG	CS IoT 2017	4	
82 R164217037 500	062417 PRACHI JAIN	CS IoT 2017	5	
	0062691 PRANJAL RAI	CS 1oT 2017	5	
84 R164217060 500	0062929 VIKRANT MALIK	CS IoT 2017	5	
	0063753 MUDRIKA TRIVEDI	CS IoT 2017	5	Dr Manoj Kumar Singh
	0062421 ABHISHEK VASHISHT	CS IT-Infra 2017	5	
	0062823 AISHWARYA CHANDRA	CS IT Infra 2017	4	
	0060144 ANANYA MALASI	CS IT Infra 2017	5	-
	0062729 NIVEDITA RAJ	CS IT Infra 2017	5	
	0063036 SHAVYA JAIN	CS IT Infra 2017	3	
	0063431 SONU PAL	CS IT Infra 2017	5	
	0062333 TEJASWA NAIK	CS IT Infra 2017	1	Dr Ravi Kiran Maddali
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	0063074 MAYUR PANDE	CS MC 2017	5	
	0062604 YASHOVARDHAN SHAKTAWAT	CS MC 2017	5	
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	0062722 ARCHIT SINGH	CS MT 2017	5	
	0062961 NISHTHA SAGAR	CS MT 2017	5	
	0063383 APARNA UPADHYAY	CS OG 2017	5	Dr R K Pavan Kumar Pannala
	0062536 ARSHAD ALI	CS OG 2017	3	DI K K rayan Kumar Pannan
	0062418 ASHISH KUMAR	CS OG 2017	2	
	0062937 AYUSH KUMAR	CS OG 2017	4	
	0062459 KARAN PRATAP SINGH	CS OG 2017	4	
	0062459 KAKAN PKATAP SINGH 0063329 SARIKA GANGWAR	CS OG 2017	4	
				De Minin I Internal
	0063050 SRISHTI AGNIHOTRI	CS OG 2017	4 5	Dr Nitin Uniyal
	0061603 VIVEK RAJ	CS OG 2017	184	
	0063881 Yuvraj Singh	CS OG 2017	5	
	0063909 Siddharth chauhan	CS OG 2017	5	
	0062444 AKSHIT CHAUHAN	CS OSS 2017	4	
	0062948 AYUSH SINGH	CS OSS 2017	3	
	0062498 GARIMA DHALL	CS OSS 2017	4	and a state of the second
	0062818 NEERAJ SINGH	CS OSS 2017	5	Dr Mukesh Kumar Singhal
	0062944 PURSHARTH RAGHUVANSHI	CS OSS 2017	3	
	0062535 SAJAL SAXENA	CS OSS 2017	4	
	0062726 SARTHAK GOYAL	CS OSS 2017	5	
	0060004 SHRISTI NEGI	CS OSS 2017	5	
117 50	0062034 TANISH JAIN	CS OSS 2017	5	Dr Amit Kumar Singh